

**I'm not a robot!**



collaborated with countless designers, developers, UXers, marketers, editors, and writers on projects of all sizes. This is good news for you: no matter what role you play, she gets what you do and knows why it's important. And, because she cares, she wants to help you understand how **content strategy can help make your life a little easier—and your end products a little more awesome.**

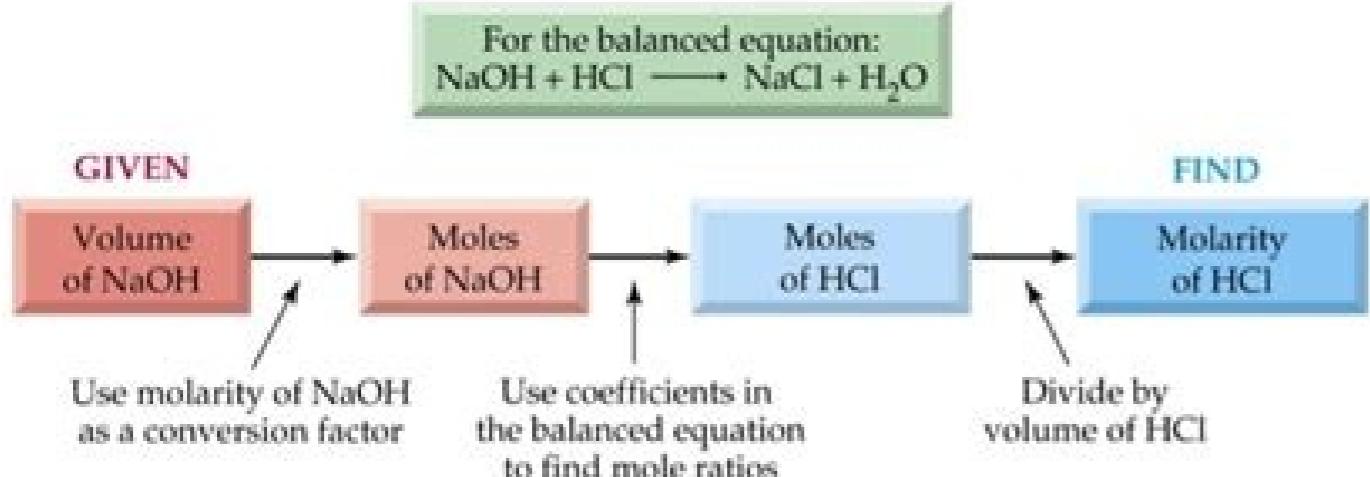
Not that long ago, I wrote an article that called upon readers to “take up the torch for content strategy.” The book you hold in your hands is that torch. So run with it. Hold it high. Be confident in your pursuit of better content. You have *The Elements of Content Strategy* to light your way.

Come on out from under the bed. We all have work to do.

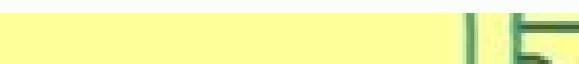
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## Solution Stoichiometry

- **Solution Stoichiometry** uses molarity as a conversion factor between volume and moles of a substance in a solution.



# TITRATION

A colorful illustration of various laboratory glassware pieces, including flasks, beakers, and test tubes, arranged in a decorative pattern. To the right, there is a detailed diagram of a titration apparatus. It features a red Erlenmeyer flask containing a red liquid on a stand. A clear glass tube with a stopcock is connected to the flask and extends upwards, ending in a smaller flask. This setup is part of a larger system with glass tubing and a small beaker at the bottom right.

# WHAT'S TITRATION?

การไทเทเรต(Titration) เป็นการวิเคราะห์หาปริมาณของสารที่ไม่ทราบความเข้มข้น ด้วยการวัดปริมาณของสารละลาย ซึ่งปริมาณของสารละลายจะมีความสัมพันธ์กับปริมาณสารโดยทำปฏิกิริยาที่ทราบปริมาณหรือความเข้มข้นที่แน่นอน

## 1 การไฟฟ้าส่องสว่าง-เบส

คือ กระบวนการนำเสนอ โดยวิธีใช้สาระโดยมาตรฐานที่ทราบ  
ค่าความเข้มข้นที่แน่นอน ให้ทำปฏิกริยากับสารตัวอย่าง โดยอาศัย  
หลักการเกิดปฏิกริยาระหว่างสารละลายกรดและเบสที่เข้าทำปฏิกริยา  
กันพอดี ทำให้คำนวณหาความเข้มข้นหรือปริมาณของสารตัวอย่าง

## วิธีการให้เกรตกรด-เบส

คือ นำสารคละลายกรดหรือเบสตัวอย่างที่ต้องการวิเคราะห์มาปิรามาณ มาทำการไห้เทรตกับสารคละลายเบสหรือกรดมาตรฐานที่ทราบค่าความเข้มข้นที่แน่นอน

ถ้าสารละลายตัวอย่างเป็นสารละลายกรด ก็ต้องใช้สารละลาย

## มาตรฐานเป็นเบส

ถ้าใช้สารละลายตัวอย่างเป็นเบส ก็ต้องใช้สารละลายมาตรฐานเป็น

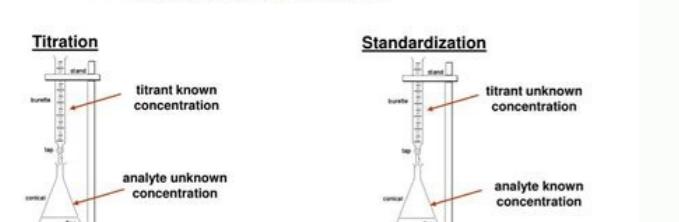
กานต์

## Titrations

## *Introduction*

- 6.) Standardization

  - Required when a non-primary titrant is used
    - Prepare titrant with approximately the desired concentration
    - Use it to titrate a primary standard
    - Determine the concentration of the titrant
    - Reverse of the normal titration process!!!





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